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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.	
10/588,008	08/01/2006	Naoto Hirosaki	KPO-002	1405	
30628 7590 08/21/2008 KANESAKA BERNER AND PARTNERS LLP 1700 DIAGONAL RD SUITE 310 ALEXANDRIA, VA 22314-2848			EXAM	EXAMINER	
			HANOR, SERENA L		
			ART UNIT	PAPER NUMBER	
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Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Application No. Applicant(s) 10/588.008 HIROSAKI ET AL. Office Action Summary Examiner Art Unit SERENA L. HANOR 1793 -- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --Period for Reply A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS. WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION. Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication. If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication - Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b). Status 1) Responsive to communication(s) filed on 01 August 2006. 2a) This action is FINAL. 2b) This action is non-final. 3) Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under Ex parte Quayle, 1935 C.D. 11, 453 O.G. 213. Disposition of Claims 4) Claim(s) 1-20 is/are pending in the application. 4a) Of the above claim(s) _____ is/are withdrawn from consideration. 5) Claim(s) 12 and 13 is/are allowed. 6) Claim(s) 1-7.10.11.14-18 and 20 is/are rejected. 7) Claim(s) 1-3,5-9 and 19 is/are objected to 8) Claim(s) _____ are subject to restriction and/or election requirement. Application Papers 9) The specification is objected to by the Examiner. 10) The drawing(s) filed on 01 August 2006 is/are: a) Accepted or b) objected to by the Examiner. Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a). Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d). 11) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152. Priority under 35 U.S.C. § 119 12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f). a) All b) Some * c) None of: Certified copies of the priority documents have been received. Certified copies of the priority documents have been received in Application No. 3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

Attachment(s)

1) Notice of References Cited (PTO-892)
2) Notice of Draftsperson's Patent Drawing Review (PTO-948)
3) Notice of Draftsperson's Patent Drawing Review (PTO-948)
4) Interview Summary (PTO-413)
Paper NoticyMail Date Paper NoticyMail Date Paper NoticyMail Date (PTO/SB/08)
Paper NoticyMail Date @@01/2006
1) Notice of Information Disclosure Statement(s) (PTO/SB/08)
6) Other:

* See the attached detailed Office action for a list of the certified copies not received.

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DETAILED ACTION

Information Disclosure Statement

i. The information disclosure statement filed on 08/01/2006 does not fully comply

with the requirements of 37 CFR 1.98(b) because: copies of the foreign patents and the

NPL document are missing. Since the submission appears to be bona fide, applicant is

given ONE (1) MONTH from the date of this notice to supply the above mentioned

omissions or corrections in the information disclosure statement. NO EXTENSION OF

THIS TIME LIMIT MAY BE GRANTED UNDER EITHER 37 CFR 1.136(a) OR (b).

Failure to timely comply with this notice will result in the above mentioned information

disclosure statement being placed in the application file with the noncomplying

information not being considered. See 37 CFR 1.97(i).

ii. The listing of references in the specification is not a proper information disclosure

statement. 37 CFR 1.98(b) requires a list of all patents, publications, or other

information submitted for consideration by the Office, and MPEP § 609.04(a) states,

"the list may not be incorporated into the specification but must be submitted in a

separate paper." Therefore, unless the references have been cited by the examiner on

form PTO-892, they have not been considered.

Claim Rejections - 35 USC § 102/103

The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

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(b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

The factual inquiries set forth in *Graham* v. *John Deere Co.*, 383 U.S. 1, 148 USPQ 459 (1966), that are applied for establishing a background for determining obviousness under 35 U.S.C. 103(a) are summarized as follows:

- Determining the scope and contents of the prior art.
- 2. Ascertaining the differences between the prior art and the claims at issue.
- Resolving the level of ordinary skill in the pertinent art.
- Considering objective evidence present in the application indicating obviousness or nonobviousness.

This application currently names joint inventors. In considering patentability of the claims under 35 U.S.C. 103(a), the examiner presumes that the subject matter of the various claims was commonly owned at the time any inventions covered therein were made absent any evidence to the contrary. Applicant is advised of the obligation under 37 CFR 1.56 to point out the inventor and invention dates of each claim that was not commonly owned at the time a later invention was made in order for the examiner to consider the applicability of 35 U.S.C. 103(c) and potential 35 U.S.C. 102(e), (f) or (g) prior art under 35 U.S.C. 103(a).

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The person having ordinary skill in the art has the capability of understanding the scientific and engineering principles applicable to the claimed invention. The references of record in this application reasonably reflect this level of skill.

1. Claims 1-7, 10, 11, 14-18 and 20 rejected under 35 U.S.C. 102(b) as anticipated by or, in the alternative, under 35 U.S.C. 103(a) as obvious over van Krevel et al. (Luminescence Properties of Terbium-, Cerium-, or Europium-Doped α -Sialon Materials) in view of Thebault et al. (U.S. Patent No. 5,411,762).

van Krevel et al. disclose a production method of an oxynitride powder, comprising applying a heat treatment in a reducing and nitriding atmosphere, to a precursor compound including nitrogen (p. 20 col. 2, *Applicants' claim* 2) and M, Si, Al, and O, where M is selected from Li, Mg, Ca, Sr, Y, La, Ce, Pr, Nd, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb, or Lu or any combination thereof, but preferably at least includes Ca or Eu (p. 20 col. 2, p. 22-23, *Applicants' claims* 17 and 18), thereby decreasing the oxygen content and increasing the nitrogen content of the precursor (*Applicants' claim* 1), in order to produce an α -sialon represented by the general formula (p. 20 col. 2, p. 22-23, *Applicants' claim* 3): $M_x Si_{12(m+n)} Al_{m+n} O_n N_{16-n}$ ($0 \le x \le 2$, $0 \le m \le 6$, and $0 \le n \le 3$).

The precursor compound is a mixture of:

a) SiX (silicon dioxide, silicon oxynitride, silicon nitride or any combination thereof),
 which turns into silicon dioxide, silicon oxynitride, or silicon nitride by heating (p. 20 col.

2, Applicants' claims 4 and 5);

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 MX (oxide, hydroxide, alkoxide, carbonate, nitrate, chloride or any combination thereof of M), which turns into an oxide, oxynitride, or nitride of M by heating (p. 20 col.

- 2. Applicants' claims 4 and 6); and
- c) AIX (oxide, hydroxide, alkoxide, carbonate, nitrate, chloride or any combination thereof of AI), which turns into aluminum oxide, aluminum oxynitride, or aluminum nitride by heating (p. 20 col. 2, Applicants' claims 4 and 7).

Said mixture is obtained by dispersing SiX particles in a solution including MX and AIX followed by drying and desolvation, wherein the M and AI compounds are attached to a surface of a SiX particle (p. 20 col. 2, *Applicants' claims 10 and 11*).

The obtained oxynitride powder is an α -sialon powder represented by the following formula (p. 21 Table 1, p. 22 col. 2, *Applicants' claim 20*):

Ca x1 Eu x2 Si 12-(m+n) Al m+n O n N 16-n, where:

 $0.4 \le x1 \le 1.5$ $0.01 \le x2 \le 0.4$ $0.8 \le m \le 3$ $0 \le n \le 2$.

"[E]ven though product-by-process claims are limited by and defined by the process, determination of patentability is based on the product itself. The patentability of a product does not depend on its method of production. If the product in the product-by-process claim is the same as or obvious from a product of the prior art, the claim is unpatentable even though the prior product was made by a different process." *In re Thorpe*, 777 F.2d 695, 698, 227 USPQ 964, 966 (Fed. Cir. 1985) (citations omitted). See MPEP 2113 [R-1]. *Applicants' claim 20*.

van Krevel et al. differs from the instant invention in that it discloses a formula of the product that falls within the instantly claimed ranges.

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It would have obvious to one of ordinary skill in the art at the time of the invention to have selected values from the instantly claimed ranges to obtain van Krevel's product, as per Applicants' claims 3, 19 and 20, because a prima facie case of obviousness exists where the claimed ranges "overlap or lie inside ranges disclosed by the prior art". In re Wertheim, 541 F.2d 257, 191 USPQ 90 (CCPA 1976); In re Woodruff, 919 F.2d 1575, 16 USPQ2d 1934 (Fed. Cir. 1990). Furthermore, "[A] prior art reference that discloses a range encompassing a somewhat narrower claimed range is sufficient to establish a prima facie case of obviousness." In re Peterson, 315 F.3d 1325. 1330. 65 USPQ2d 1379. 1382-83 (Fed. Cir. 2003). See MPEP 2144.05 [R-5].

van Krevel et al. differs from the instant invention in that it does not specifically disclose the M and Al compounds as being attached to a surface of a SiX particle.

It would have obvious to one of ordinary skill in the art at the time of the invention to have known that the M and Al compounds would be attached to the surfaces of the SiX particles, as per Applicants' claims 10 and 11, because it is known that the particles in powder form are in the form of a matrix with SiX particles being the host in the matrix, thus the known definition of a sialon powder. "[T]he discovery of a previously unappreciated property of a prior art composition, or of a scientific explanation for the prior art's functioning, does not render the old composition patentably new to the discoverer." Atlas Powder Co. v. Ireco Inc., 190 F.3d 1342, 1347, 51 USPQ2d 1943, 1947 (Fed. Cir. 1999). See MPEP 2112 [R-3] I.

van Krevel et al. differs from the instant invention in that it does not disclose the order of dispersing the SiX particles in a solution including the MX and AlX particles.

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It would have obvious to one of ordinary skill in the art at the time of the invention to have mixed the particles by dispersing the SiX particles in a solution including the MX and AIX particles, as per Applicants' claim 11, because the selection of any order of mixing ingredients is prima facie obvious. See Ex parte Rubin, 128 USPQ 440 (Bd. App. 1959), In re Burhans, 154 F.2d 690, 69 USPQ 330 (CCPA 1946), and In re Gibson, 39 F.2d 975, 5 USPQ 230 (CCPA 1930). See MPEP 2144.04 [R-6] IV C.

van Krevel et al. differs from the instant invention in that it discloses only removing the solvent from the mixture of SiX, MX, and AlX, rather than specifically subjecting the mixture to drying and desolvation.

It would have obvious to one of ordinary skill in the art at the time of the invention to have recognized that the step of removing the solvent involves drying and/or desolvation, as per Applicants' claim 11, because "[e]xpress suggestion to substitute one equivalent for another need not be present to render such substitution obvious." In re Fout. 675 F.2d 301, 213 USPQ 532 (CCPA 1982). See MPEP 2143 B Example 1.

Thebault et al. disclose a process of making a sialon-based material by applying a heat treatment in a reducing and nitriding atmosphere (col. 3 lines 5-15), which includes at least an ammonia gas (col. 1 lines 66-68, col. 2 lines 9-14 and 43-47, Applicants' claim 14) or a mixed gas atmosphere of ammonia and hydrocarbon (col. 2 lines 43-47, Applicants' claim 15), wherein the hydrocarbon gas is methane or propane (col. 2 lines 43-52, Applicants' claim 16).

It would have been obvious to one of ordinary skill in the art at the time of the invention to have modified the process of van Krevel et al. by using an atmosphere

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comprising ammonia gas and a hydrocarbon gas such as methane, as per Thebault et al. (col. 1 lines 66-68, col. 2 lines 9-14 and 43-52, col. 3 lines 5-15), as per Applicants' claims 14-16, because of the stated advantage that the presence of a gaseous carbon compound in a nitrogen gas atmosphere, or ammonia, since it is known that ammonia gas decomposes to nitrogen gas, is necessary for obtaining a product comprising sialon (Thebault et al., col. 2 lines 55-58), as per Applicants' claim 1.

 Claims 1-7, 10, 11, 14-18 and 20 are rejected under 35 U.S.C. 102(b) as anticipated by or, in the alternative, under 35 U.S.C. 103(a) as obvious over Kohtoku et al. (U.S. Patent No. 4,845,059) in view of Thebault et al. (U.S. Patent No. 5,411,762).

Kohtoku et al. disclose a production method of an oxynitride powder, comprising applying a heat treatment in a reducing and nitriding atmosphere, which includes at least an ammonia gas (col. 4 lines 24-45, *Applicants' claim 14*), to a precursor compound including nitrogen (col. 2 lines 25-45, *Applicants' claim 2*) and M, Si, Al, and O, where M is selected from Li, Mg, Ca, Sr, Y, La, Ce, Pr, Nd, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb, or Lu or any combination thereof, but preferably at least includes Ca or Eu (col. 2 lines 47-64, *Applicants' claims 17 and 18*), thereby decreasing the oxygen content and increasing the nitrogen content of the precursor (*Applicants' claim 1*) in order to produce an α -sialon represented by the general formula (col. 3 lines 26-40, *Applicants' claim 3*): M_x Si $_{12 < (m+n)}$ Al $_{m+n}$ O $_n$ N $_{16-n}$ ($0 \le x \le 2$, $0 < m \le 6$, and $0 \le n \le 3$).

The precursor compound is a mixture of:

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a) SiX (silicon dioxide, silicon oxynitride, silicon nitride or any combination thereof),
 which turns into silicon dioxide, silicon oxynitride, or silicon nitride by heating (col. 2
 lines 1-36 and 65-68, Applicants' claims 4 and 5);

- b) MX (oxide, hydroxide, alkoxide, carbonate, nitrate, chloride or any combination thereof of M), which turns into an oxide, oxynitride, or nitride of M by heating (col. 2 lines 1-21 and 37-64, *Applicants' claims 4 and 6*); and
- c) AIX (oxide, hydroxide, alkoxide, carbonate, nitrate, chloride or any combination thereof of AI), which turns into aluminum oxide, aluminum oxynitride, or aluminum nitride by heating (col. 2 lines 1-45, col. 3 lines 1-8, Applicants' claims 4 and 7).

Said mixture is obtained by dispersing SiX particles in a solution including MX and AIX followed by drying and desolvation, wherein the M and AI compounds are attached to a surface of a SiX particle (col. 4 lines 6-11, Applicants' claims 10 and 11).

The obtained oxynitride powder is an α -sialon powder represented by the following formula (col. 3 lines 26-40. *Applicants' claim 20*):

Ca x1 Eu x2 Si 12-(m+n) Al m+n O n N 16-n, where:

 $0.4 \le x1 \le 1.5$ $0.01 \le x2 \le 0.4$ $0.8 \le m \le 3$ $0 \le n \le 2$.

"[E]ven though product-by-process claims are limited by and defined by the process, determination of patentability is based on the product itself. The patentability of a product does not depend on its method of production. If the product in the product-by-process claim is the same as or obvious from a product of the prior art, the claim is unpatentable even though the prior product was made by a different process." In re-

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Thorpe, 777 F.2d 695, 698, 227 USPQ 964, 966 (Fed. Cir. 1985) (citations omitted). See MPEP 2113 [R-1]. Applicants' claim 20.

Kohtoku et al. differs from the instant invention in that the ranges of x, x1, x2, m, and n are overlapping and/or lie within the disclosed ranges of the instant invention.

It would have obvious to one of ordinary skill in the art at the time of the invention to have selected values from Kohtoku et al. also disclosed by the instant invention, as per Applicants' claims 3, 19 and 20, because a prima facie case of obviousness exists in the case where the claimed ranges "overlap or lie inside ranges disclosed by the prior art". In re Wertheim, 541 F.2d 257, 191 USPQ 90 (CCPA 1976); In re Woodruff, 919 F.2d 1575, 16 USPQ2d 1934 (Fed. Cir. 1990). Furthermore, "[A] prior art reference that discloses a range encompassing a somewhat narrower claimed range is sufficient to establish a prima facie case of obviousness." In re Peterson, 315 F.3d 1325, 1330, 65 USPQ2d 1379, 1382-83 (Fed. Cir. 2003). See MPEP 2144.05 [R-5].

Kohtoku et al. differs from the instant invention in that it does not specifically disclose the M and Al compounds as being attached to a surface of a SiX particle.

It would have obvious to one of ordinary skill in the art at the time of the invention to have known that the M and Al compounds would be attached to the surfaces of the SiX particles, as per Applicants' claims 10 and 11, because it is known that the particles in powder form are in the form of a matrix with SiX particles being the host in the matrix, thus the known definition of a sialon powder. "[T]he discovery of a previously unappreciated property of a prior art composition, or of a scientific explanation for the prior art's functioning, does not render the old composition patentably new to the

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discoverer." Atlas Powder Co. v. Ireco Inc., 190 F.3d 1342, 1347, 51 USPQ2d 1943, 1947 (Fed. Cir. 1999). See MPEP 2112 [R-3] I.

Kohtoku et al. differs from the instant invention in that it does not disclose the order of mixing SiX, MX, and AIX in a solution.

It would have obvious to one of ordinary skill in the art at the time of the invention to have selectively dispersed the SiX into a solution of a mixture of MX and AlX, as per Applicants' claim 11, because the selection of any order of mixing ingredients is prima facie obvious. See Ex parte Rubin, 128 USPQ 440 (Bd. App. 1959), In re Burhans, 154 F.2d 690, 69 USPQ 330 (CCPA 1946), and In re Gibson, 39 F.2d 975, 5 USPQ 230 (CCPA 1930). See MPEP 2144.04 [R-6] IV C.

Kohtoku et al. differs from the instant invention in that it discloses simply removing the solvent from the mixture of SiX, MX, and AIX.

It would have obvious to one of ordinary skill in the art at the time of the invention to have recognized that the step of removing the solvent involves drying and/or desolvation, as per Applicants' claim 11, because "[e]xpress suggestion to substitute one equivalent for another need not be present to render such substitution obvious." In re Fout, 675 F.2d 301, 213 USPQ 532 (CCPA 1982). See MPEP 2143 B Example 1.

Thebault et al. disclose a process of making a sialon-based material by applying a heat treatment in a reducing and nitriding atmosphere (col. 3 lines 5-15), which includes at least an ammonia gas (col. 1 lines 66-68, col. 2 lines 9-14 and 43-47, Applicants' claim 14) or a mixed gas atmosphere of ammonia and hydrocarbon (col. 2

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lines 43-47, Applicants' claim 15), wherein the hydrocarbon gas is methane or propane (col. 2 lines 43-52, Applicants' claim 16).

It would have been obvious to one of ordinary skill in the art at the time of the invention to have modified the process of van Krevel et al. by using an atmosphere comprising ammonia gas and a hydrocarbon gas such as methane, as per Thebault et al. (col. 1 lines 66-68, col. 2 lines 9-14 and 43-52, col. 3 lines 5-15), as per Applicants' claims 14-16, because of the stated advantage that the presence of a gaseous carbon compound in a nitrogen gas atmosphere, or ammonia, since it is known that ammonia gas decomposes to nitrogen gas, is necessary for obtaining a product comprising sialon (Thebault et al., col. 2 lines 55-58), as per Applicants' claim 1.

 Claims 1-7, 10, 14-18 and 20 are rejected under 35 U.S.C. 102(b) as anticipated by or, in the alternative, under 35 U.S.C. 103(a) as obvious over Mitomo et al. (U.S. Patent No. 6,632,379 B2) in view of Thebault et al. (U.S. Patent No. 5,411,762).

Mitomo et al. disclose a production method of an oxynitride powder, comprising the step of applying a heat treatment in a reducing and nitriding atmosphere (col. 7 lines 9-17 and lines 52-57), to a precursor compound including nitrogen (col. 7-8, *Applicants' claim* 2) and M, Si, Al, and O, where M is selected from Li, Mg, Ca, Sr, Y, La, Ce, Pr, Nd, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb, or Lu or any combination thereof, but preferably at least includes Ca or Eu (col. 2 lines 48-59, col. 3 lines 21-30, *Applicants' claims* 17 and 18), thereby decreasing the oxygen content and increasing the nitrogen content of the precursor (col. 5 lines 1-3, *Applicants' claim* 1) in order to produce an α-sialon represented by the general formula (col. 2 lines 48-59, *Applicants' claim* 3):

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$$M_x Si_{12-(m+n)} Al_{m+n} O_n N_{16-n}$$
 $(0 \le x \le 2, 0 \le m \le 6, \text{ and } 0 \le n \le 3).$

The precursor compound is a mixture of:

- a) SiX (silicon dioxide, silicon oxynitride, silicon nitride or any combination thereof),
 which turns into silicon dioxide, silicon oxynitride, or silicon nitride by heating (col. 7 lines 9-13. col. 8. Applicants' claims 4 and 5):
- b) MX (an oxide, hydroxide, alkoxide, carbonate, nitrate, chloride or any combination thereof of M), which turns into an oxide, oxynitride, or nitride of M by heating (col. 5 lines 4-9, col. 7 lines 9-13, col. 8, *Applicants' claims 4 and 6*); and
- c) AIX (an oxide, hydroxide, alkoxide, carbonate, nitrate, chloride or any combination thereof of AI), which turns into aluminum oxide, aluminum oxynitride, or aluminum nitride by heating (col. 7 lines 9-13, col. 8, Applicants' claims 4 and 7). Furthermore, MX and AIX are attached to a surface of the SiX (Applicants' claim 10).

The obtained oxynitride powder is an α -sialon powder represented by the following formula (col. 2 lines 48-59, col. 3 lines 21-30, col. 8 lines 19-47, *Applicants'* claim 20): Ca $_{x1}$ Eu $_{x2}$ Si $_{12\text{-(m+n)}}$ Al $_{m+n}$ O $_{n}$ N $_{16-n}$, where:

 $0.4 \le x1 \le 1.5$ $0.01 \le x2 \le 0.4$ $0.8 \le m \le 3$ $0 \le n \le 2$.

"[E]ven though product-by-process claims are limited by and defined by the process, determination of patentability is based on the product itself. The patentability of a product does not depend on its method of production. If the product in the product-by-process claim is the same as or obvious from a product of the prior art, the claim is unpatentable even though the prior product was made by a different process." In re-

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Thorpe, 777 F.2d 695, 698, 227 USPQ 964, 966 (Fed. Cir. 1985) (citations omitted). See MPEP 2113 IR-11. Applicants' claim 20.

Mitomo et al. differs from the instant invention in that the ranges of x, x1, x2, m, and n are overlapping and/or lie within the disclosed ranges of the instant invention.

It would have obvious to one of ordinary skill in the art at the time of the invention to have selected values from Mitomo et al. also disclosed by the instant invention, as per Applicants' claims 3, 19 and 20, because a prima facie case of obviousness exists in the case where the claimed ranges "overlap or lie inside ranges disclosed by the prior art". In re Wertheim, 541 F.2d 257, 191 USPQ 90 (CCPA 1976); In re Woodruff, 919 F.2d 1575, 16 USPQ2d 1934 (Fed. Cir. 1990). Furthermore, "[A] prior art reference that discloses a range encompassing a somewhat narrower claimed range is sufficient to establish a prima facie case of obviousness." In re Peterson, 315 F.3d 1325, 1330, 65 USPQ2d 1379, 1382-83 (Fed. Cir. 2003). See MPEP 2144.05 [R-5].

Mitomo et al. differs from the instant invention in that it does not specifically disclose the M and Al compounds as being attached to a surface of a SiX particle.

It would have obvious to one of ordinary skill in the art at the time of the invention to have known that the M and Al compounds would be attached to the surfaces of the SiX particles, as per Applicants' claim 10, because it is known that the particles in powder form are in the form of a matrix with SiX particles being the host in the matrix, thus the known definition of a sialon powder. "[T]he discovery of a previously unappreciated property of a prior art composition, or of a scientific explanation for the prior art's functioning, does not render the old composition patentably new to the

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discoverer." Atlas Powder Co. v. Ireco Inc., 190 F.3d 1342, 1347, 51 USPQ2d 1943, 1947 (Fed. Cir. 1999). See MPEP 2112 IR-3I I.

Thebault et al. disclose a process of making a sialon-based material by applying a heat treatment in a reducing and nitriding atmosphere (col. 3 lines 5-15), which includes at least an ammonia gas (col. 1 lines 66-68, col. 2 lines 9-14 and 43-47, Applicants' claim 14) or a mixed gas atmosphere of ammonia and hydrocarbon (col. 2 lines 43-47, Applicants' claim 15), wherein the hydrocarbon gas is methane or propane (col. 2 lines 43-52, Applicants' claim 16).

It would have been obvious to one of ordinary skill in the art at the time of the invention to have modified the process of van Krevel et al. by using an atmosphere comprising ammonia gas and a hydrocarbon gas such as methane, as per Thebault et al. (col. 1 lines 66-68, col. 2 lines 9-14 and 43-52, col. 3 lines 5-15), as per Applicants' claims 14-16, because of the stated advantage that the presence of a gaseous carbon compound in a nitrogen gas atmosphere, or ammonia, since it is known that ammonia gas decomposes to nitrogen gas, is necessary for obtaining a product comprising sialon (Thebault et al., col. 2 lines 55-58), as per Applicants' claim 1.

Conclusion

Claims 1-7, 10, 11, 14-18 and 20 have been rejected.

Claims 1-3, 5-9 and 19 have been objected to.

Claims 8, 9, 12, 13 and 19 have not been rejected under either 35 U.S.C. 102 or 35 U.S.C. 103 because the limitations of these claims are not taught in the reference(s) of record.

Any inquiry concerning this communication or earlier communications from the

examiner should be directed to SERENA L. HANOR whose telephone number is

(571)270-3593. The examiner can normally be reached on Monday - Thursday 8:00

AM - 5:30 PM EST.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's

supervisor, Stanley Silverman can be reached on (571) 272-1358. The fax phone

number for the organization where this application or proceeding is assigned is 571-

273-8300.

Information regarding the status of an application may be obtained from the

Patent Application Information Retrieval (PAIR) system. Status information for

published applications may be obtained from either Private PAIR or Public PAIR.

Status information for unpublished applications is available through Private PAIR only.

For more information about the PAIR system, see http://pair-direct.uspto.gov. Should

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Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a

USPTO Customer Service Representative or access to the automated information

system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

SLH 08/14/08

/Timothy C Vanoy/

Primary Examiner, Art Unit 1793